



# Cork: Its Origin and Industrial Uses

*Gilbert E. Stecher*

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"[...]which forms small granular crystals very soluble in boiling water, alcohol and ether. It fuses at 300 degrees Fahrenheit and sublimes in acidular crystals. It is also produced when nitric acid acts on stearic, margaric or oleic acid. The chemical analysis is given as (C<sub>8</sub>H<sub>14</sub>O<sub>4</sub>) and I am inclined to believe it is the truer one, as it is much later than Fownes'. This suberic acid has been further broken up to ascertain its fundamental characteristics and it was found to partake of the two compounds suberone and suberate. Suberone (C<sub>14</sub>H<sub>24</sub>O<sub>2</sub>) being regarded as the ketone of suberic acid, an aromatic liquid compound obtained when suberic acid is distilled with an excess of lime.[24] Also described as a colorless oil with an odor of peppermint and a boiling[...]".

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